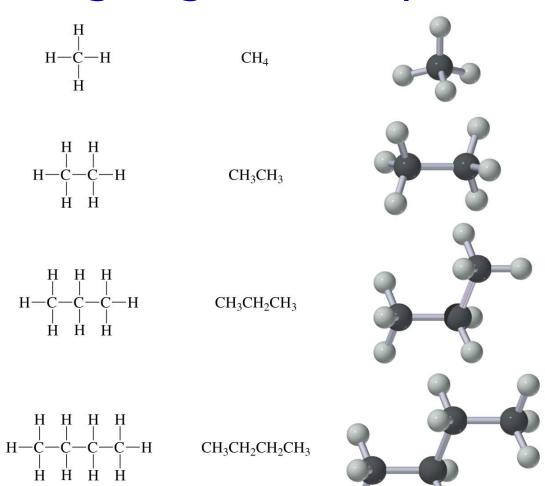
Chapter 4: Organic Compounds

Part 1: Organic Compounds and Alkanes



$$H_3C$$
 C
 C
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

IUPAC Nomenclature

→ All compounds are named as derivatives of the longest single carbon chain.

$$\rightarrow$$

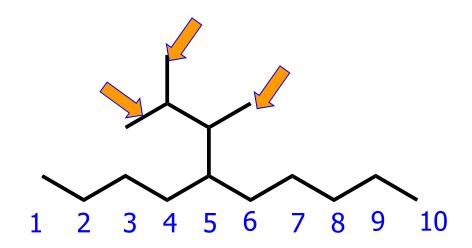
→The modifying prefixes such as di, tri, tetra, etc. are used when you have more than one of the same group.

Functional group nomenclature

The prefixes fluoro, chloro, bromo, iodo used to indicate presence of halogens (halo).

They are treated in same manner as alkyl substituents.

- → More complex appendages are named as derivatives of the longest carbon chain in a side group.
- →These are enclosed in parentheses.



Common Branched Groups:

$$H_3C$$
 H_3C
 H_3C

Rules for IUPAC Nomenclature

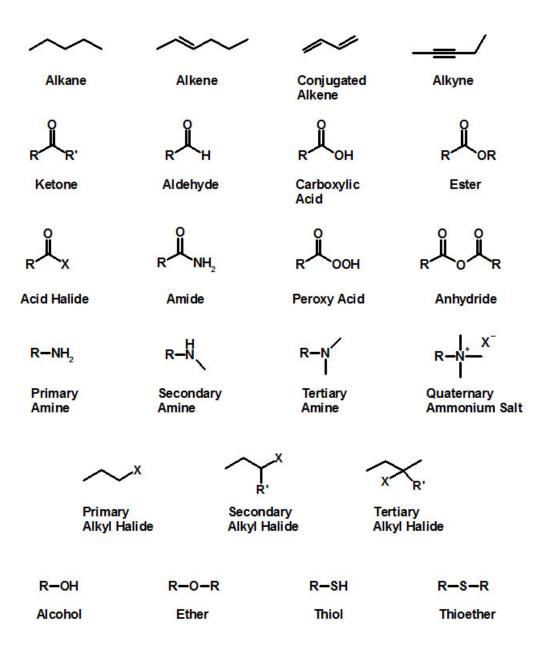
- → Many organic compounds named by changing the suffix.
- → New suffix dominates #'s over alkane substituents

 \rightarrow

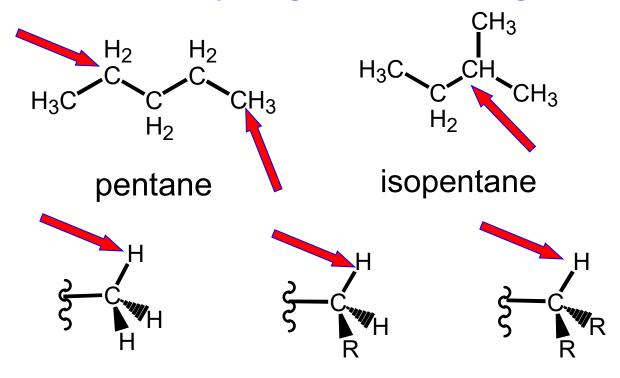


 NH_2

Major Organic Chemistry Functional Groups

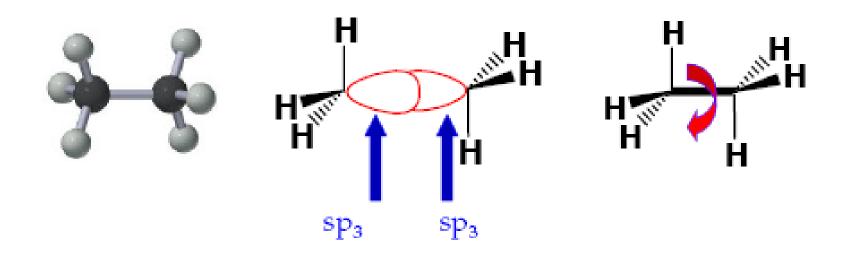


We need to make a distinction between different types of Carbons, Hydrogens, or Nitrogens

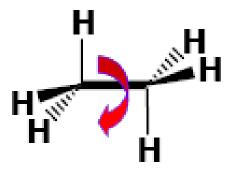


Conformations of Alkanes

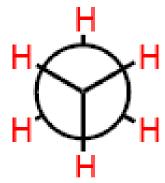
Let's look again at ethane....

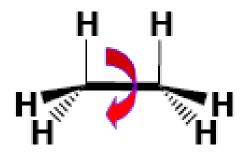


Conformational Isomers

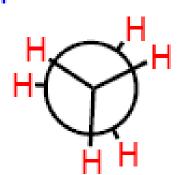


staggered conformer

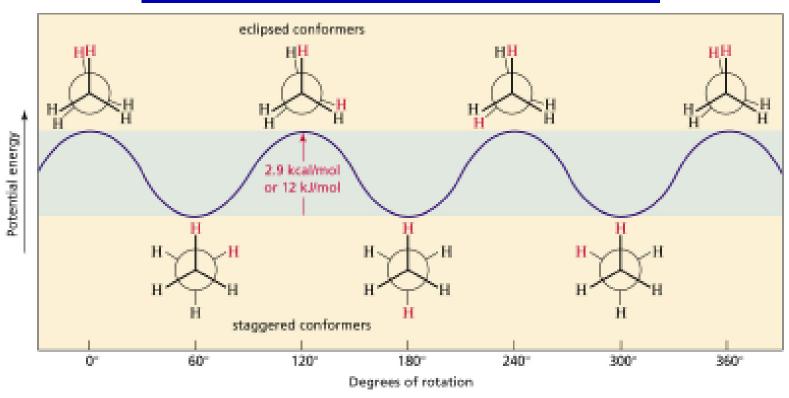




eclipsed conformer



Conformations of Ethane



For Next Time....

- ▶ Monday More Chapter 4 (4.1 4.7)
 - BRING YOUR MODEL SET!
- Homework Problems Chapter 3#1,4,7,15,34,35,37,39,43,44, 47
- Homework Problems Chapter 4
- #1, 6, 10, 19, 25, 28, 36, 43, 48, 51,52, 63