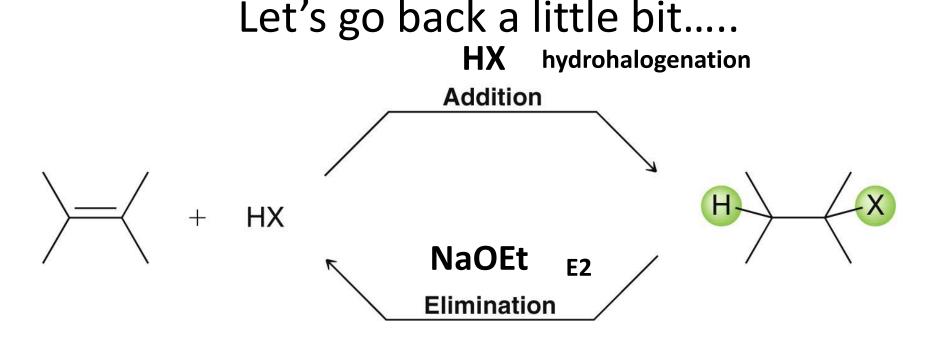
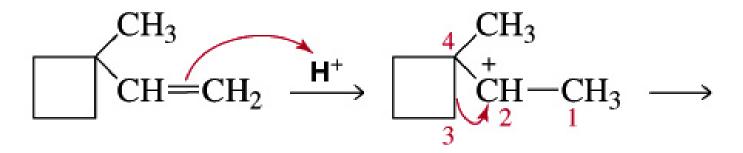
## Chapter 8 part 2:

#### Acid Catalyzed Addition of Water

- Today (8.4 8.7) Addition reactions Acid Catalyzed Hydration/Oxymercuration-Reduction
- Friday (8.7-8.10) Hydrogenation
   Halogenation/Halohydrin formation



## **Ring Expansion**



2<sup>0</sup> carbocation

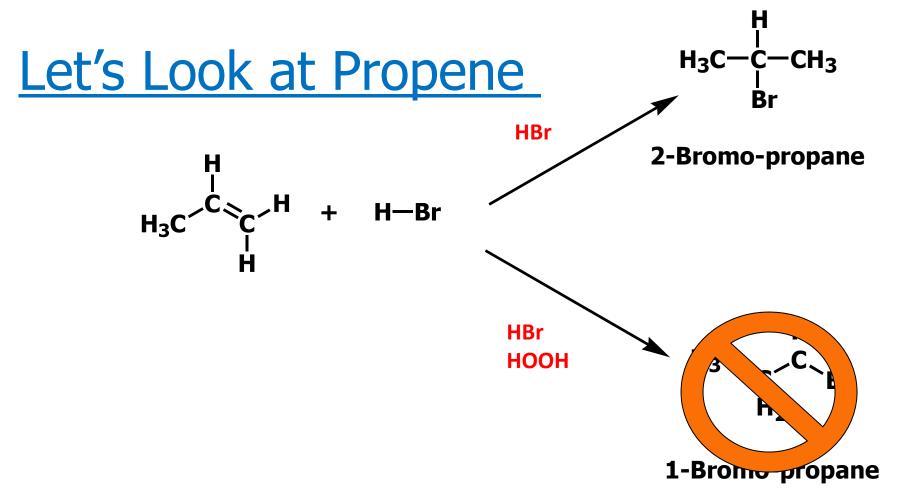
$$\begin{array}{c|c}
CH_3 & CH_3 \\
 & \xrightarrow{4} \\
 & \xrightarrow{2} \\
 & \xrightarrow{1}
\end{array}$$

$$\begin{array}{c}
CH_3 \\
Br \\
 & \xrightarrow{-} \\
 & CH_3
\end{array}$$

$$\begin{array}{c}
CH_3 \\
 & \xrightarrow{-} \\
 & CH_3
\end{array}$$

a more stable 30 carbocation

## **Hydrohalogenation – Regiochemistry**



- ▶ This is a called a <u>regioselective</u> reaction.
- ▶ In a <u>regioselective</u> reaction, one constitutional isomer is the major or the only product (8.3...)

# <u>Addition of Water</u>

- Hydration of an alkene is the addition of H-OH to to give an alcohol
- Acid catalysts are used in high temperature industrial processes: ethylene is converted to ethanol

Ethylene

### Electrophilic Addition to Alkenes

This is called <u>hydrohalogenation</u>.

$$H_3C$$
  $\longrightarrow$   $H_3C$   $\longrightarrow$   $H_3C$   $\longrightarrow$   $H_3C$ 

1-methylcyclohexene

This is called acid-catalyzed <u>hydration</u>.

## **Acid Catalyzed Hydration**

• The components of water (-H and -OH) are added across a C=C double bond.

### Addition of Water to an Alkene

$$H_3C$$
  $+$   $H_3C$   $+$   $H_3C$   $+$   $H_3C$ 

1-methylcyclohexene

1-methylcyclohexanol

To make an *alcohol* 

#### Addition of an Alcohol to an Alkene

$$H_3C$$
  $+$   $H_3C$   $O$   $H$   $H_3C$   $H_3C$ 

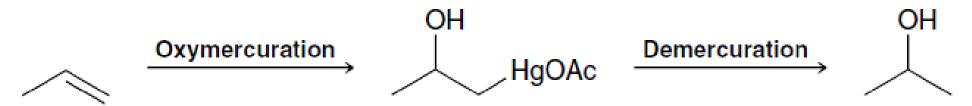
1-methylcyclohexene

1-methoxy-1-methylcyclohexane

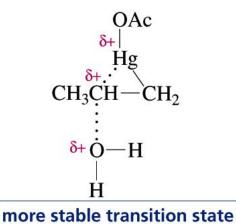
To make an **ETHER** 

## Oxymercuration-Demercuration

- Because rearrangements often produce a mixture of products, the synthetic utility of Markovnikov hydration reactions is somewhat limited.
- OXYMERCURATION-DEMERCURATION is an alternative process that can yield Markovnikov products more cleanly.



# Oxymercuration-reduction of an Alkene



OAc

$$\begin{array}{c}
\delta+|\\
Hg\\
-\vdots\\
CH_3CH-CH_2\\
\vdots\\
\delta+O-H\\
|\\
H
\end{array}$$

less stable transition state

#### For Next Time....

Suggested Homework Problems Chapter 8

# 1, 2, 5, 9, 12,13, 18, 24, 27, 31, 42-46, 52, 57,62,63